1505

A Convenient One-pot Synthesis of Polyenic Enol Ethers

Yen Vo-Quang,* Denis Carniato, Liliane Vo-Quang, and François Le Goffic

Laboratoire de Bioorganique et Biotechnologies, E.N.S.C.P.–C.E.R.C.O.A., 11, rue Pierre et Marie Curie, 75231 Paris Cedex 05, France

Diethyl alkoxymethylphosphonates react with unsaturated aldehydes to give polyenic enol ethers with a good stereoselectivity.

Polyene synthesis by Wittig-type reactions has been well documented.¹ The high stereoselectivity and experimental simplicity of the phosphonate olefin synthesis² have made it a popular method for the introduction of unsaturation concurrently with chain lengthening, particularly in the field of natural products and their analogues.^{1–3}

This synthetic tool seemed particularly useful for the synthesis of the polyenic enol ethers required for our studies on the synthesis of a mutagen and its analogues. Several phosphonate reagents have been developed for the preparation of enol ethers from carbonyl compounds,⁴⁻⁶ but unfortunately their use has presented some problems.^{7,8} A diphenylphosphine oxide reagent has recently been reported to give satisfactory yields of methyl enol ethers.^{9,10}

The present paper describes a convenient and highly stereoselective method for the synthesis of the polyenic enol ethers (3) by the reaction of the polyenal (2) with a carbanion of the diethyl alkoxymethylphosphonate (1). Several methods are available for the extension of double bond conjugation in systems containing an aldehyde function.⁴ Two of these were selected as being the most appropriate for the present work.¹¹



The diethyl alkoxymethylphosphonates (1) (R = Me, Et, PhCH₂, MeOCH₂CH₂OCH₂) were readily prepared by the Michaelis-Arbuzov reaction of the chloromethyl ethers with triethylphosphite in 85–95% yields. Their anions [lithium di-isopropylamine (LDA) (1.5 equiv.), tetrahydrofuran-hexane, -80 to -100 °C, argon] when added to the polyenals (2), gave good yields of the isolated adducts (3) (80–90%).†

The adducts (3), when treated with KOBu^t or KH (2 equiv., 50 °C, 15 min), gave the polyenic enol ethers (4) (15–25% yield of distilled product).† In a one-pot procedure, the lithiated adduct of (1) was condensed with (2) and the reaction mixture was heated to reflux and left for ≥ 4 h at this temperature. A yield of distilled product of *ca*. 45–50% was obtained† [b.p. °C/mmHg, R = PhCH₂, n = 0 98/0.05, n = 1 118/0.05, n = 2 162/0.05; R = Et, n = 2 76/0.05; R = Me, n = 2 98/0.5; R = MeOCH₂CH₂OCH₂, n = 2 (40% yield of crude oil)].

This synthesis of the unsaturated polyenic enol ethers is highly stereoselective: the newly-formed double bond has predominantly the *E* configuration ($n = 0 \ E: Z \ 80: 20; n = 1, 2 \ E: Z \ 100:0$; determined by ¹H n.m.r. spectroscopy). The structures of the vinyl ethers were evident from their ¹H n.m.r. spectra; the vinyl protons next to oxygen resonate at low field, and the *Z* (7 Hz) and *E* (13 Hz) ³*J* coupling constants were unambiguous.^{9,12}

It is of note that, in preliminary experiments using diphenylbenzyloxyphosphine oxide, prepared according to the method of Warren *et al.*,⁹ the ratio of the geometrical isomers of (4) was 50: 50 for n = 0. Preliminary results, obtained with Li–Zn

† Reaction conditions were not optimized. Satisfactory i.r. spectra were obtained for all products. For (3, n = 0), ¹H n.m.r. (100 MHz, CDCl₃-Me₄Si, assignment by spin decoupling and [P] irradiation), δ 0.92 (2 × t, 3H, Me), 1.32 (t, 6H, OCH₂CH₃), 1.60 (m, 2H, CH₂), 3.05 (br. s, OH, exch. D₂O), 3.75 (m, 2H, CH-O and CH-P), 4.18 (m, 4H, OCH₂CH₃), 4.72 (m, 2H, CH₂Ph), 7.35 (s, 5H, Ph); corresponding spectra for (3, n = 1, 2)are similar. For E-(4, n = 0), ¹H n.m.r. [conditions as for (3)], δ 1.05 (t, 3H, Me), 2.85 (m, 2H, CH₂), 4.70 (s, 2H, CH₂Ph), 4.90 and 6.31 (both dt, 1H each, ³J 12.5 Hz, vinyl H), 7.35 (s, 5H, Ph); spectra for [4, n = 0 (Z), 1, 2] are analogous. exchange in a one-pot procedure, seem promising but the reaction is less stereoselective.

We thank the N.M.R. Spectroscopy Laboratory of the E.N.S.C.P. and in particular M. J. Pouet for technical assistance and discussions.

Received, 25th July 1983; Com. 992

References

- J. I. G. Cadogan, 'Organophosphorus Reagents in Organic Synthesis,' Academic Press, London, 1979; H. J. Bestmann, *Pure Appl. Chem.*, 1980, **52**, 771.
- 2 W. S. Wadsworth, Jr., 'Synthetic Applications of Phosphoryl Stabilized Anions in Organic Reactions,' vol. 25, Wiley, 1977, pp. 73-253; J. Boutagy and R. Thomas, *Chem. Rev.*, 1974, 87.
- P. J. Van Den Tempel and H. O. Huisman, Tetrahedron, 1966, 22, 293; J. A. Edwards, V. Schwarz, J. Fatkos, M. L. Maddox, and J. H. Fried, J. Chem. Soc., Chem. Commun., 1971, 292; H. De Koning, G. N. Mallo, A. Springer-Fidder, K. E. C. Subramanian-Ehrart, and H. O. Huisman, Recl. Trav. Chim. Pays-Bas, 1973, 92, 683; A. E. Asato, D. Mead, M. Denny, T. T. Bopp, and R. S. H. Liu, J. Am. Chem. Soc., 1982, 104, 4979; K. C. Nicolaou, N. A. Petasis, R. E. Zipkin, and J. Uenishi, *ibid.*, p. 5555.
- 4 S. F. Martin, Synthesis, 1979, 633.
- 5 J. C. Gilbert and U. Weerasooriya, J. Org. Chem., 1983, 48, 448.
- 6 A. F. Kluge and I. S. Clousdale, J. Org. Chem., 1979, 44, 4847.
- 7 G. Lavielle and D. Reisdorg, C.R. Acad. Sci., Ser. C, 1971, 272, 100; G. Sturtz, G. Lavielle, and H. Normant, Chem.-Ztg., 1972, 96, 503.
- 8 M. Green, J. Chem. Soc., 1963, 1324.
- 9 C. Earnshaw, C. J. Wallis, and S. Warren, J. Chem. Soc., Perkin Trans. 1, 1979, 3099.
- 10 M. Schlosser and H. B. Tuong, Chimia, 1976, 30, 197.
- D. Marshall and M. C. Whiting, J. Chem. Soc., 1956, 4081;
 W. R. Roush and A. G. Myers, J. Org. Chem., 1981, 46, 1511;
 F. Bohlmann and H. Jastrow, Chem. Ber., 1962, 95, 2939;
 B. M. Mikhailov and L. V. Povarov, Z. Obshch. Khim., 1959, 29, 2079;
 D. Fishman, J. T. Klug, and A. Shani, Synthesis, 1981, 137.
- 12 V. Subramanyam, E. H. Silver, and A. H. Saloway, J. Org. Chem., 1976, 41, 1272.